

How Does Cotton Get into the Lipstick?

Dr. Gabriele Kaiser



Introduction

Whether in Chanel, Dior, Estée Lauder, Babor, Lancôme or Douglas, hydrogenated cottonseed oil is a substance that is increasingly found on the list of ingredients for decorative cosmetics and personal care products. Cottonseed oil is extracted from the seeds of the cotton plant [1] and is valued as an edible oil in many countries. Since the cotton plant contains a natural toxin against insect feeding, the oil must first be refined and the harmful gossypol must be removed. This results in a pale yellow liquid with a high content of unsaturated fatty acids and vitamin E.

Due to its high stability, hydrogenated cottonseed oil is frequently used in cosmetic products. The term hydrogenation describes the addition of hydrogen to unsaturated double bonds in the presence of a catalyst and is also referred to as "hardening". Through hydrogenation, the light yellow oil turns into a white or almost white powder. This process, however, usually leaves some unsaturated bonds. Therefore, in addition to saturated fat, hydrogenated cottonseed oil usually still contains about 2% unsaturated fatty acids [2].

As a cosmetic ingredient, hydrogenated cottonseed oil features moisturizing properties and a non-greasy texture; it leaves the skin feeling smooth and soft [3]. It can be found in skin cleansing products, lip liners, eyeliners and lipsticks, among others.

Melting and Crystallization Behavior

Like all oils and fats, hydrogenated cottonseed oil belongs to the lipid group and consists of triglycerides of various fatty acids, among them palmitic acid and stearic acid. The melting range of lipids depends on many different factors such as chain length, chain branching, number of double bonds, degree of esterification, and arrangement in the crystal structure [4], since fats and oils can exist in different polymorphic forms or modifications.

The NETZSCH DSC 300 *Caliris® Classic* was employed for the investigations detailed here. By virtue of its small footprint, it fits into (almost) every laboratory. The measurement parameter are summarized in table 1.

Table 1 Measurement parameters	
Instrument	DSC 300 Caliris® Classic
Sample	Hydrogenated cottonseed oil
Sample weight	approx. 6 mg
Crucible	Al, closed, cold-welded
Temperature range	0°C 90°C
Heating/cooling rates	2, 5, 10 and 20 K/min
Atmosphere	N_2



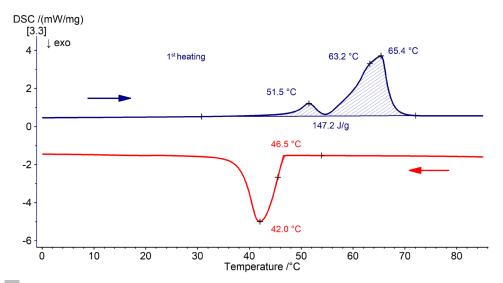
Measurement Results

In the current case, the sample shows a broad melting range between approx. 40°C and 72°C upon heating (fig. 1, blue curve).

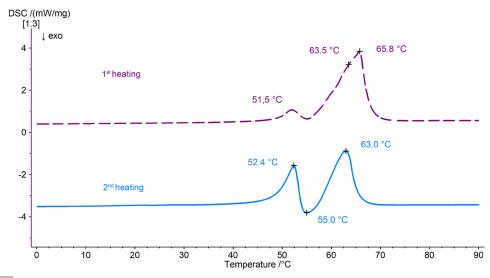
Several endothermic effects can be seen in this temperature range: the most significant are at about 52°C, 63°C and 65°C (peak temperature in each case).

During the subsequent controlled cooling (red curve in fig. 1), the substance starts to crystallize at approx. 47°C. The solidification effect is not structured.

If the sample is heated a second time after cooling (again at a heating rate of 10 K/min, light blue curve in fig. 2), a completely different picture is obtained than in the 1st heating, reflecting the polymorphic character of the hydrogenated cottonseed oil. Along with two distinct endothermal effects at 52°C and 63°C (peak temperature in each case), an exothermic effect occurs in between at approx. 55°C (also peak temperature). The temperature position of the endothermic effect at 52°C (light blue curve in fig. 2) agrees well with the corresponding endothermic effect in the 1st heating (dashed violet curve). The second endothermic peak seems to have shifted slightly to the left as compared to the 1st heating.



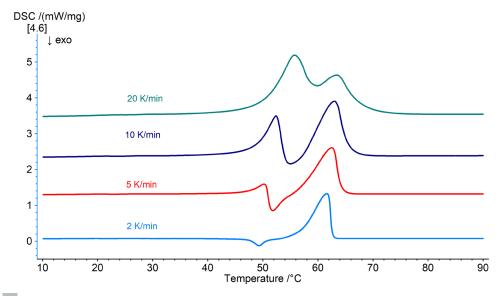
1 DSC measurement on hydrogenated cottonseed oil, 1st heating and subsequent cooling; sample mass: 6.1 mg, heating/cooling rates: 10 K/min



2 DSC measurement on hydrogenated cottonseed oil, 1st and 2nd heating runs; sample mass: 6.2 mg, heating rate: 10 K/min.



APPLICATIONNOTE How Does Cotton Get into the Lipstick?



DSC measurements on hydrogenated cottonseed oil, multiple heating runs; sample masses: 6.0 to 6.3 mg, heating rates: 2 to 20 K/min; individual scaling.

By varying the heating rate during the 2nd heating, it is possible to completely suppress the first endothermic effect and to separate the exothermic peak from the second endothermic effect at low heating rates (2 K/min, light blue curve in figure 3). At higher heating rates (5, 10 or 20 K/min), the first endothermal effect occurs and becomes more and more dominant with increasing heating rate until the exotherm is entirely compensated at a heating rate of 20 K/min.

It is therefore possible that the exothermic peak in the range from 50°C to 55°C is based on a structural change. Further investigations using such techniques as X-ray structural analysis would be necessary to verify this hypothesis.

Conclusion

Hydrogenated cottonseed oil is a hydrogenated vegetable oil that can be used in cosmetics and creams as an alternative [5]. Its rather complex melting behavior can be described phenomenologically in a fast and straightforward manner by means of the DSC 300 *Caliris® Classic*.

Literature

- [1] https://www.cosmeticsinfo.org/ingredients/ hydrogenated-cottonseed-oil/
- [2] https://en.wikipedia.org/wiki/Cottonseed_oil
- [3] https://www.100percentpure.com/pages/ingredient-hydrogenated-cottonseed-oil#:~:text=ln%20addition%20to%20its%20moisturizing,suitable%20for%20all%20skin%20types.
- [4] C. Reitz, PhD thesis, Extrudierte Fettmatrizes mit retardierter Wirkstofffreigabe, Heinrich-Heine-Universität Düsseldorf, 2007, pp 11-13
- [5] https://file.wuxuwang.com/hpe/HPE6/HPE6_326.pdf

